Synthesis and Diels Alder Reactions of (+)-(S)-1-t-Butylsulfonyl-1-p-Tolylsulfinylethene, a New Masked Chiral Ketene Equivalent

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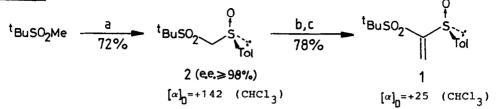
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Abstract Enantiomerically pure (+)-(S)s-1-t-buty|sulfony|-1-p-toly|sulfiny|ethene (1) was easily prepared in three steps from t-buty| methyl sulfone. The catalyzed Diels-Alder reactions of 1 with cyclopentadiene were highly stereoselective. The major adduct was transformed into <math>(+)-(1R,4R)-dehydronorcamphor and (+)-(1R,4S)-2-t-buty|sulfony|-2,5-norbornadiene.

Enantiomerically pure a, β -unsaturated sulfoxides have currently received a considerable attention as dienophiles in asymmetric Diels-Alder reactions¹. To the best of our knowledge, only three of them, (+)-(R)-ptolyl vinyl sulfoxide², (+)-(S,S)-1,1-bis(p-tolylsulfinyl)ethene³ and (+)-(R)-ethynyl p-tolylsulfoxide⁴, have been used as masked chiral ketene equivalents However their utility is very limited due to their moderate reactivity and facial stereoselectivity. In this communication we report the synthesis and some Diels-Alder reactions of a new chiral ketene equivalent: (+)-(S)-1-t-butylsulfonyl-1-p-tolylsulfinylethene (1).

Scheme 1



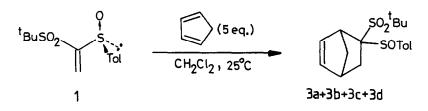
a: "BuLi, THF, -78°C, 30 min; then (S)s-p-tol-SO(OMent) (0.5 eq.), r.t., 5h. b' Me2NH (2 eq), CH2O (2 eq), MeOH, 0°C, 3h. c. IMe, CaCO3, CH3CN, r.t., 3h.

Unsaturated sulfoxide 1 was easily prepared following the sequence of reactions shown in scheme 1. Deprotonation of t-butyl methyl sulfone with n-BuLi and sulfinylation with (-)-(S)s-menthyl p-toluenesulfinate⁵ gave (+)-(S)-t-butylsulfonyl p-tolylsulfinyl methane (2), which was easily

purified by silica gel chromatography⁶ (72% yield). The optical yield of compound (S)-2 (ee \ge 98%) has been determined by ¹H-NMR using Yb(hfc)₃ as chiral shift reagent, while the (S)-configuration has been proposed taking into account that the Andersen reaction takes place with complete inversion at the sulfur center⁷. Treatment of (S)-2 with dimethylamine and formaldehyde in methanol gave a 1:1 mixture of the diastereomeric ß-dimethylamino Mannich products, which were directly submitted to cuaternization and elimination of the amine molety (IMe/CaCO₃) yielding crude (S)-1-t-butylsulfonyl-1-p-tolylsulfinylethene (1)⁶ (78% yield).

The results corresponding to the Diels-Alder reaction of 1 with cyclopentadiene are summarized in scheme 2.

Scheme 2

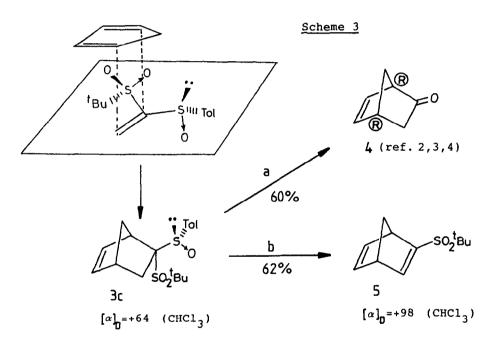


Entry	Lewis acid	Time(h)	Y1eld 3(%)ª	3a:3b:3c:3db	Yield 3c(%) ^c
1		66	364	9:23:35.23	
2	ZnBr₂ (0.8eq.)	6	68	12::88	58
3	Eu(fod)3 (1.0eq)) 42	70	8: 92	62
4	S102•	48	72	10:84 6	63

a: Yield of pure adducts 3 after chromatography. b Isomer ratio determined by $^{1}H-NMR$. c: Yield of pure adduct 3c after chromatography. d: 60% of (S)-1 was recovered. e: Carried out without solvent⁹ (weight ratio SiO₂/1=10:1).

In the absence of catalysts the reaction of 1 with cyclopentadiene is very slow (36% of conversion after 66 h at room temperature) and poorly stereoselective (entry 1), while in the presence of strong Lewis acids (BF3 Et20, Et2ClAl or EtCl2Al) there is no cycloaddition due to the faster decomposition of dienophile 1. Fortunately, in the presence of moderate catalysts such as $ZnBr_2$ (entry 2), $Eu(fod)_3$ (entry 3) and $S1O_2^3$ (entry 4) the Diels-Alder reaction took place mildly, giving good yields of Moreover these catalyzed cycloadditions occur with adducts 3. high (one diastereomer clearly predominates). The major stereoselectivity adduct 3c was easily separated from the mixture by flash chromatography¹⁰ (58-63% y1eld).

Scheme 3 shows two alternatives for the elimination of the sulfingl chiral auxiliary. Hydrolysis of 3c with TiCl₃ (15% in water) in acetic acid gave the known (+)-(1R,4R)-dehydronorcamphor^{2,3,4} (4), showing the utility of (S)-1 as a latent chiral ketene equivalent. On the other hand, the pyrolytic elimination was easily effected by heating 3c at 70°C in toluene containing 2.0 equiv. of trimethylphosphite. (+)-(1R,4S)-2-tbutylsulfonyl-2,5-norbornadiene (5) was obtained in 62% yield after silica gel chromatography¹¹. Its enantiomeric purity (ee>80%), has been determined by ¹H-NMR using Yb(hfc)₃ as quiral shift reagent.



a: TICl3 (15% in water), AcOH, r.t., 5d; b P(OMe)3 (2eq.), toluene, 70°C, 4d.

Obtaining (1R,4R)-dehydronocamphor by hydrolysis of 3c unambigously shows the 1R,4R configuration of this major adduct. Since Diels-Alder reactions of cyclopentadiene with 1,1-disubstituted olefins usually result in a predominance of the adduct with the bulkier substituent in *endo* configuration¹² (in order to avoid the steric destabilization with the CH₂ fragment on diene), we propose for 3c the configuration 1R,2R,4R shown in scheme 3. This stereochemistry would come from the *endo* approach of diene, with respect to the t-butylsulfonyl group, to the less hindered face of the S-*cis* conformation of dienophile 1¹³.

REFERENCES AND NOTES

1.- Posner G.H , "*The Chemistry of Sulfones and Sulfoxides*" Ed. Patai, S., Rappoport, Z., Stirling, C, John Wiley & Sons, New York (1988), Chapter 16, p. 845 and references cited therein

2 - Maignan C., Raphael R A., Tetrahedron, 1983, 39, 3245.

3.- Araı Y., Kuwayama S.; Takeuchı Y., Koızumı T., Synth. Commun, 1986, 16, 233.

4 - Maignan C.; Belkasmioui F., Tetrahedron Lett., 1988, 29, 2823.

5.- Solladie G., Hutt J.; Girardin A., Synthesis, 1987, 173.

6.-(±)-2 was prepared in the same way by using (±) methyl p-toluenesulfinate instead of (-)-(S)-menthyl p-toluenesulfinate (76% yield, m p.=88-89°C) Data for (+)-(S)-2. m.p.=116-117°, [α]p=143 (CHCl₃, c=1). ¹H NMR (CDCl₃) δ 7.70 and 7 36 (AA'BB' system, 4H, arom), 4.45 and 4.19 (AB system, 2H, CH₂, J=13.6 Hz), 2.44 (s, 3H, CH₃) and 1.47 (s, 9H, ^tBu).

7.- The same method has been previously applied to the synthesis of (+)-(S)-phenylsulfonyl p-tolylsulfinyl methane (Annunziata R.; Cinquini M; Cozzi F, Synthesis, 1979, 535).

8.- Crude dienophile (+)-(S)-1 was obtained as an oil with a chemical purity >95% ([a] $_D$ =+25, CHCl₃, c=1). On the contrary (±)-1 (obtained from (±)-2) is a solid (m.p. 116-118°C) ¹H-NMR (CDCl₃) δ : 7.64 and 7 30 (AA'BB' system, 4H, arom), 7.20 (d, 1H, J=2.0 Hz, C=CH), 6.83 (d, 1H, J=2.0 Hz, C=CH), 2.40 (s, 3H, CH₃) and 1.44 (s, 9H, ^tBu).

9.- Veselovsky V, Gybin A., Lozanova A.; Moiseenkovy A.; Smit W., Tetrahedron Lett., 1988, 29, 175.

10 - Data for (+)-3c: m.p. 109-110°C, Rr=0.35 (hexane/ethyl acetate 3:1), [a]_0= +64 (CHCl_3, c=1). E.e.>80% determined by ¹H-NMR with 0.6 eq. of Yb(hfc)_3. ¹H-NMR (CDCl_3) δ ⁻ 7.64 and 7.34 (AA'BB' system, 4H, arom), 6.44 (dd, 1H, J=5 7 and 3.1 Hz, C=CsH), 6.06(dd, 1H, J=5.7 and 2.7 Hz, C=CsH), 3.65(m, 1H, C1H), 2.65 (m, 1H, C4H), 2.43 (s, 3H, CH_3), 2.03 (dd, 1H, J=14.3 and 3.2 Hz, C_3H), 1.68 (s, 9H, ^tBu), 1.66 (d, 1H, J=5.1 Hz, C7H), 1.54 (dd, 1H, J=12.3 and 3.6 Hz, C_3H) and 1.42 (d, 1H, J=5.1 Hz, C7H).

11.- ¹H-NMR of 5 (CDCl₃) δ . 7.70 (d, 1H, J=3.5 Hz, C=C₃H), 6.98 (dd, 1H, J=5.1 and 2.9 Hz, C=C₆H), 6.83 (dd, 1H, J=5.1 and 3.2 Hz, C=C₅H), 3.93 (m, 1H, C₄H), 3.83 (m, 1H, C₁H), 3.58 (dt, 1H, J=6.66 and 1.5 Hz, C₇H), 2.26 (m, 1H, C₇H) and 1.31 (s, 9H, ^tBu).

12.~ a) Angell E.C.; Fringuelli F.; Minuti L.; Tatichi A.; Wenkert E., J. Org. Chem., 1988, 53, 4325. b) Alonso I.; Carretero J.C.; Garcia Ruano J.L., Tetrahedron Lett., 1989, 30, 3853. c) Rebiere F.; Riant O.; Kagan H.B., Tetrahedron Asymmetry, 1990, 1, 199.

13.- The greater stability of the *S-cis* conformation with respect to the *S-trans* in related 1,1-disubstituted vinylsulfoxides has been reported (Koizumi T., Aray Y., Takayama H , *Tetrahedron Lett.*, 1987, *28*, 3689).